

Determination of Technetium by Laser Induced Photoacoustic Spectroscopy Coupled with a Wave-Length Shifter Method

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Abstract

Laser induced photoacoustic spectroscopy (LPAS) was applied to the determination of ^{99}Tc in a solution, coupled with quantitative complexation of Tc by thiocyanate ion (NCS^-) to form $[\text{Tc}(\text{IV})(\text{NCS})_6]^{2-}$, which had an intense absorption peak at 500 nm. Sensitivity of the quantitative detection could be improved down to a concentration of $10^{-9} \text{ mol dm}^{-3}$ which was one order of magnitude better than that for thiourea complex.

Introduction

The long-lived radioactive nuclide, ^{99}Tc , is one of the most important fission products with high fission yields (6.06%). Its long half life ($2.13 \times 10^5 \text{ y}$) and characteristic pure β^- emission make its determination difficult. The authors have tried to develop a specific determination by $^{99}\text{Tc}(\gamma, \gamma')^{99\text{m}}\text{Tc}$ nuclear excitation [1–3], as well as by laser induced photoacoustic spectroscopy (LPAS) coupled with quantitative a complexation of technetium with thiourea (tu) to form $\text{Tc}(\text{tu})_6^{3+}$ [4]. In the case of the LPAS method using $\text{Tc}(\text{tu})_6^{3+}$, which has an intense absorption peak at 488 nm with a molar extinction coefficient of $6400 \text{ dm}^3 \text{ mol}^{-1} \text{ cm}^{-1}$ [4], a detection limit of ^{99}Tc could be obtained down to the order of $10^{-8} \text{ mol dm}^{-3}$.

In this paper, the hexakis(thiocyanato)technetate(IV) ion ($[\text{Tc}(\text{NCS})_6]^{2-}$) has been chosen to improve the determination limit, because its absorption peak has a molar extinction coefficient larger than that for $\text{Tc}(\text{tu})_6^{3+}$. This work has been undertaken to provide a basis for development of a new chemical speciation using LPAS.

Experimental

The LPAS system consists of a nitrogen gas laser (Laserphotonics Co., UV24), a dye laser (Laserphotonics Co., DL14), and a gated boxcar averager system

(Stanford Research Systems, SR250). The detection part which includes piezo-electric transducers (PZT) (Tohoku Kinzoku Co.) was home-made, and operation details was given in a previous report [4].

A 2 ml sample solution in a quartz cell (1 cm path length) was submitted to LPAS measurements by excitation with monochromatic laser pulse. The wave length of the excitation beam was adjusted to 500 nm using Coumarin-481 dye material.

$[\text{Tc}(\text{NCS})_6]^{2-}$ was prepared by the method according to Crouthamel [6]. A solution of 1 M NH_4SCN was added into a 4 M H_2SO_4 solution containing pertechnetate. Within 5 min, during stirring, the color of the solution changed to red-violet. Then a 1.5 times larger volume of acetone was added to stabilize the complex. An aliquot of this stock solution was diluted with distilled water to give an appropriate concentration of technetium for the LPAS measurements. An optical absorption spectrum was measured with a spectrophotometer (Hitachi 50–100). The radioactivity of Tc was determined with a liquid scintillation counter (Aloka LSC-700).

Results and discussion

The molar extinction coefficient of $[\text{Tc}(\text{NCS})_6]^{2-}$ at 500 nm was obtained as $46900 \text{ dm}^3 \text{ mol}^{-1} \text{ cm}^{-1}$ by optical absorption spectrophotometry, which was in good agreement with the reported value [6] in the presence of acetone. Fig. 1 shows change of absorbance at 500 nm after mixing a pertechnetate solution with 1 M NH_4SCN in the presence of acetone. It confirmed quantitative formation of $[\text{Tc}(\text{NCS})_6]^{2-}$ and its stability was guaranteed for standing of up to one day. In the absence of acetone, the absorption spectrum gradually changed to give another technetium species which has an absorption maximum at 410 nm, as shown in Fig. 2. This species could be attributed to $[\text{Tc}(\text{III})(\text{NCS})_6]^{3-}$ produced by further reduction of $[\text{Tc}(\text{IV})(\text{NCS})_6]^{2-}$ [7, 8].

When the acetone containing solution was submitted to the LPAS measurement, gas evolution could be observed and the true photoacoustic signals were drastically disturbed by this cavitation. Degasification

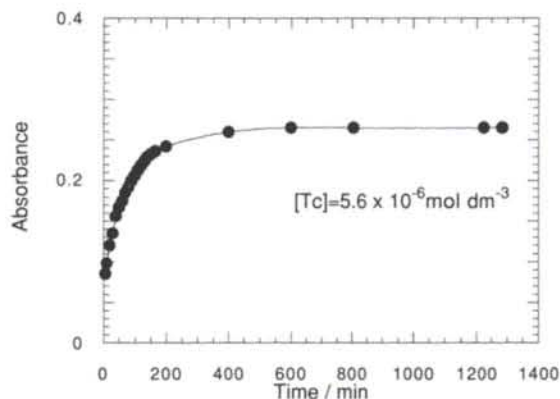


Fig. 1. The formation of $[\text{Tc}(\text{NCS})_6]^{2-}$ after mixing pertechnetate and 1 M NH_4SCN solution in the presence of acetone.

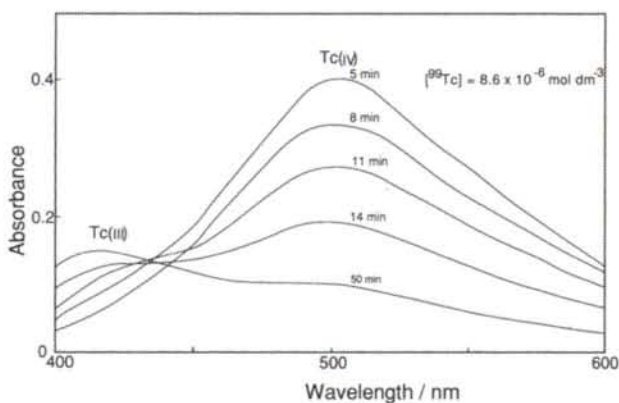


Fig. 2. The changes in absorption spectra of $[\text{Tc}(\text{NCS})_6]^{2-}$ in the absence of acetone.

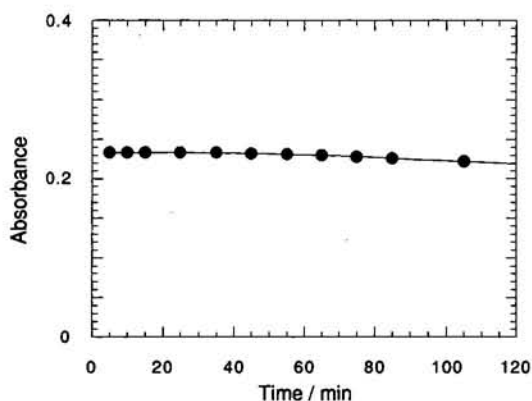


Fig. 3. The time dependence of absorbance of $[\text{Tc}(\text{NCS})_6]^{2-}$ when diluted with distilled water.

of the solution by ultrasonic treatment was not effective. Gas generation was not observed in the absence of acetone. Therefore, an aliquot of the stock solution of $[\text{Tc}(\text{NCS})_6]^{2-}$ stabilized by acetone was diluted with distilled water, and LPAS measurements were made immediately. The absorbance of the diluted solution at 500 nm slowly decreased with increasing time, as shown in Fig. 3. This change was slow enough to allow a quick measurement with 3 min after the sample

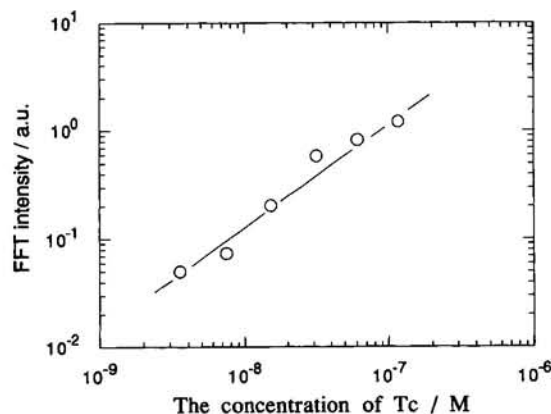


Fig. 4. The normalized FFT peak intensity as a function of the concentration of ^{99}Tc .

preparation by increasing the laser pulse repetition to 20 Hz and by repeating 3 data averagings. Fast Fourier Transfer (FFT) analysis was used for the averaged data to evaluate peak intensity [4].

The pressure wave, p , which is related to the intensity of photoacoustic signal, can be expressed by [9],

$$p = \text{const. } \beta V_a E \alpha / C_p \quad (1)$$

where β is a thermal expansion coefficient, E is pulsed laser energy, C_p is specific heat at constant pressure, V_a is acoustic velocity, and α is optical absorbance. α must be the value for a solute, ϵC (ϵ : molar extinction coefficient, C : concentration) in a given medium. The net photoacoustic intensity of the solute can be obtained by subtracting the photoacoustic signal of the medium itself. Thus, the net acoustic pressure normalized to E can be directly related to the concentration of the solute, $[\text{Tc}(\text{NCS})_6]^{2-}$. In order to perform LPAS measurements for various concentrations of $[\text{Tc}(\text{NCS})_6]^{2-}$, different volumes of the stock solution containing acetone were diluted with distilled water to give a constant volume of 2 ml. This resulted in different compositions of water and acetone for each sample, which varies β and C_p in Equation (1). Acetone mixing would cause an increased background, because the increase of β and decrease of C_p are expected by addition of acetone. However, this effect was sufficiently small in the case of the reference solutions which changed the concentration of acetone up to 8%. Almost constant photoacoustic signals were obtained, irrespective of change in the acetone concentration.

The FFT peak intensity normalized to the laser power is plotted against the concentration of ^{99}Tc in Fig. 4. A linear relationship could be obtained over the concentration range 10^{-7} – 10^{-9} mol dm^{-3} and the determination limit was improved by one order of magnitude better than the case using thiourea complex.

Conclusion

The determination limit was improved down to the order of 10^{-9} mol dm^{-3} ^{99}Tc ; 2 ml of this solution is

equivalent to about 440 mBq of ^{99}Tc . This value is comparable to that obtained by activation analysis [1].

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